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### Chapter 7 Holt Chemistry Test

Holt Chemistry Chapter 7 The Mole and Chemical Composition. The unit used to measure the amount of...  $6.022 \times 10^{23}$  of anything. the mass in grams of one mole of a subs... the periodic table - an elements mass l... mole.

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Chapter 7 Holt Chemistry. STUDY. PLAY. What does the term coordination number in ionic crystals refer to. The # of oppositely charged ions surrounding a particular ion. How many electrons does nitrogen gain in order to achieve a noble gas electron configuration. 3.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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Holt Chemistry Chapter 15 - Section 15.1 -15.3 Holt Chemistry Chapter 15 - Sections 15.1 -15.3 An acid that ionizes completely in aqueous solution an acid that is only slightly ionized in aqueous solution A base that dissociates completely into ions in solution.

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