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Problems In your

notebook, solve the

following problems. ...

Use the Table 2.1 to

answer the following

questions. 6.

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METALLIC BONDING
Practice Problems In
your notebook, answer
the following SECTION
7.1 IONS For each
element below, state
(i) the number of
valence electrons in
the atom, (ii) the
electron dot structure,
and (iii) the chemical
symbol(s) for the most

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Resources ____ 12.

The molecular
structure of carbon
dioxide is one carbon
atom with two oxygen
atoms on opposite
sides of it. ____ 13.

Covalent bonds exist
when combining atoms
give up or accept
electrons. ____ 14. A
molecule contains two
atoms. Part C Matching

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Match each description
in Column B to the
correct term in Column
A.

12

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MOLECULAR

COMPOUNDS 8

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Radioactive decay a. is
a spontaneous process.

b. occurs when an
unstable nucleus loses
energy. c. does not

require any input of

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energy. d. all of the
above _____ 13. Which
of the following types
of radiation has no
mass and no charge?
a. alpha c. gamma b.
beta d. positron _____

14. When U-238 _____
undergoes radioactive
decay by losing an
alpha particle,

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your notebook, answer
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element below, state
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the atom, (ii) the
electron dot structure,
and (iii) the chemical
symbol(s) for the most

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Test

Write your answer in the space provided.
Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that,

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perhaps, the grass is
dying.

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