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Modern Chemistry Chapter 21 Section

an unstable nucleus that undergoes radioactive decay. alpha particle. 2 protons and 2 neutrons bound together and is emitted from the nucleus during some kind of radioactive decay. beta particle. an electron emitted from the nucleus during some kind of radioactive decay. positron. particle that has the same mass as an electron, but has a positive charge and is emitted during some kind of radioactive decay.

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Modern Chemistry 84 . Name Section Quiz, continued Class Date 5. In some instances, the concentration of a solution is expressed as molality instead of molarity because a. molality is easier to calculate. b. molarity applies only to solid-liquid solutions.

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Modern Chemistry 111 Chapter Test Name Class Date Chapter Test B, continued 18. Dissolution processes with negative enthalpies of solution are processes. 19. is the solution process with water as the solvent. 20. Liquid solutes and solvents that are not soluble in each other are. 21. A solute molecule that is surrounded by solvent molecules is ...

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SECTION 2 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances. a. temperature increases, volume stays the same

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Modern Chemistry 30 Chapter Test Name Class Date Chapter Test A, continued ____ 21. According to Einstein, which of the following can behave like a wave and also like a stream of particles? a. a noble gas b. the atomic nucleus c. electromagnetic radiation d. a hydrogen atom in the ground state ____ 22.

Assessment Chapter Test A

MODERN CHEMISTRY HRW materiã eopvrighžed under appearing this work, CHAPTER 22 TEST 85 . Name CHAPTER 22 TEST continued Date Class 11. The spontaneous disintegration of a nucleus into a slightly less massive and more stable nucleus with emission of particles and electromagnetic radiation is

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Modern Chemistry 6 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. ____ When an electron is added to a neutral atom, energy is (a) always absorbed.

Modern Chemistry Chapter 5 Review Answers Section 1

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When a molecular compound dissolves and ionizes in a polar solvent, ions are formed. 21. In a strong electrolyte, all or almost all of the dissolved compound exists as ions in aqueous solution. In a weak electrolyte, little of the dissolved compound exists as ions in aqueous solution. 22.

Assessment Chapter Test B

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