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Stoichiometry Example C 6H 6 + Br 2-----> C 6H 5Br + HBr Benzene (C 6H 6) reacts with Bromine to produce bromobenzene (C 6H 6Br) and hydrobromic acid. If 30. g of benzene reacts with 65 g of bromine and produces 56.7 g of bromobenzene, what is the percent yield of the reaction? 30.g 65 g 56.7 g ...

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